## Chemical Structure of Deoxynupharidine, the Alkaloid Obtained from Nuphar Japonicum

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Hofmann degradation and subsequent ozonolysis of deoxynupharidine, C<sub>15</sub>H<sub>23</sub>NO (I), the alkaloid obtained from the roots of Nupher Japonicum afforded 3-furaldehyde (Dimedone derivative, m.p. 167-170°, Found: C, 70.35; H, 7.18. Calcd. for  $C_{21}H_{26}O_5$ : C, 70.39; H, 7.26), which was identified by silver oxide oxidation to 3-furoic acid (m. p. 122-123°. Found: C, 53.89; H, 3.47. Calcd. for  $C_5H_4O_3$ : C, 53.57; H, 3.57), and characterization of the latter by means of mixed melting point and infrared spectrum. Compound I which gives an orange colored Ehrlich reaction, has a  $\lambda_{max}$  at  $213 \text{ m}\mu$  (log  $\varepsilon=3.9$ ), and shows infrared absorptions at 873, 1500, and 3120 cm<sup>-1</sup> corresponding to a furan ring; these absorption maxima and coloration dissappear upon conversion into tetrahydrodeoxynupharidine by catalytic reduction of I.

Palladium dehydrogenation of I, or dehydrogenation of  $C_{15}H_{24}NO \cdot CH_3$ , (II) (Perchlorate, m. p. 97–98°) resulting from Hofmann degradation and catalytic reduction of (I), afforded  $C_{15}H_{19}ON$  (III) (Platinum chloride complex, m. p. 155–160°); potassium permanganate oxidation of (III) yielded pyridine 2,5-dicarboxylic acid (Dimethylester, m. p. 163–164°. Found: C, 55.36; H, 4.39; N, 7.24. Calcd. for  $C_9H_9NO_4$ : C, 55.33;

H, 4.56; N, 7.11), which was identified through a mixed melting point test and infrared absorption comparison with an authentic sample.

Further Hofmann degradation and ozonolysis of II gave a dicarboxylic acid, m. p.  $84.5-85^{\circ}$  (Found: C, 52.78: N, 7.22. Calcd. for  $C_7H_{12}O_4$ : C, 52.50; H, 7.50.), which was shown to be  $\alpha$ -methyladipic acid from the infrared spectra of its dimethyl ester and p-bromophenancyl ester (measured in chloroform).

An acid, m. p.  $181-183^{\circ}$  (Found: C, 68.07; H, 9.85; N, 6.53. Calcd. for  $C_{12}H_{21}NO_2$ : C, 68.25; H, 10.02; N, 6.64.), resulting from the nitric and chromic acid oxidation of I, and the two  $pK_a'$  values at 2.0 and 10.8 suggested that it is presumably an  $\alpha$ -amino acid of formula IV.

The mentioned facts point to the proposed structure I, and the reactions involved may be represented by the following scheme:

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